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Domino Knoevenagel hetero-Diels–Alder reactions: a stereoselective synthesis of sugar fused furo[3,2-*b*]pyrano[4,3-*d*]pyran derivatives

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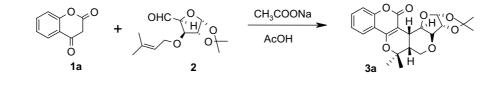
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Abstract—The *O*-prenyl derivative of a sugar aldehyde derived from D-glucose undergoes smooth intramolecular domino Knoevenagel hetero-Diels–Alder reactions with 1,3-diones to afford a novel class of carbohydrate analogues, *cis*-fused furopyranopyrans in good yields with a high degree of chemoselectivity. The stereochemistry of the products was assigned by NMR. © 2004 Published by Elsevier Ltd.

Coumarin derivatives are widely distributed in Nature and are reported to have a wide range of biological activities such as anti-coagulant, insecticidal, anthelmintic, hypnotic and anti-fungal activity. Others are phytoalexins and are inhibitors of HIV protease.^{1,2} Many naturally occurring compounds such as isoethuliacoumarin A, isoethuliacoumarin B, isoethuliacoumarin C, ethuliacoumarin A, ethuliacoumarin B and pterophyllin possess the pyrano[3,2-*c*]coumarin skeleton and have been isolated from various sources.3,4 The domino Knoevenagel intramolecular hetero-Diels-Alder reaction is one of the most powerful synthetic routes for the synthesis of various heterocycles and natural products.^{5,6} However, there are no examples of domino Knoevenagel hetero-Diels-Alder reactions using an O-prenylated sugar aldehyde derived from D-glucose.

In this article, we describe a novel protocol for the synthesis of sugar fused furo[3,2-b]pyrano[4,3-d]pyrans via domino Knoevenagel hetero-Diels–Alder reactions between an *O*-prenylated sugar aldehyde and 1,3-diones. Thus, treatment of 4-hydroxycoumarin with an *O*-prenyl derivative of a sugar aldehyde in the presence of sodium acetate in acetic acid at 80 °C resulted in the formation of *cis*-fused pyrano[3,2-c] coumarin **3a** in 82% yield (Scheme 1).

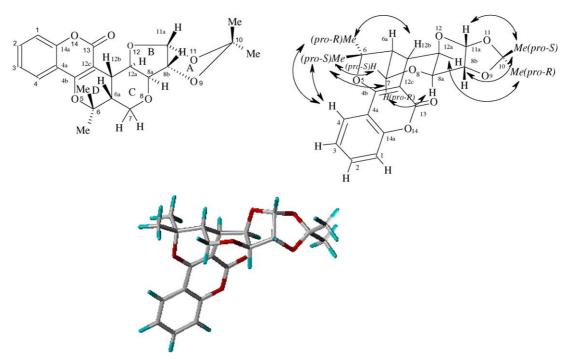
The reaction proceeds via a tandem Knoevenagel and hetero-Diels–Alder pathway. This reaction is a highly stereoselective affording exclusively *cis*-fused pyrano[3,2-*c*]coumarin derivatives. The *cis*-stereochemistry of the products was assigned by detailed NMR studies. (¹H NMR studies of product **3a** were carried out in CDCl₃ solution at 500 MHz). The couplings of ${}^{3}J_{\text{H6a-H7pro-}R}$ =



Scheme 1.

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The structure of 3a, characteristic NOE's and the energy minimised structure.

11.7 Hz, ${}^{3}J_{\text{H6a-H7(pro-S)}} = 4.3$ Hz, ${}^{3}J_{\text{H8a-H12a}} = 2.3$ Hz, ${}^{3}J_{\text{H6a-H12b}} = 5.5$ Hz, and ${}^{3}J_{\text{H12a-H12b}} = 2.3$ Hz, and a NOE between H_{7(pro-R)}-H_{8a} amply support a ${}^{6a}C_{8a}$ conformation for the pyran ring (C). The observation of a long range ω -coupling ${}^{3}J_{\text{H7(pro-S)-H12b}} = 1.3$ Hz further supports this chair conformation. The small ${}^{3}J_{\text{H6a-H12b}} = 5.5$ Hz value shows the *cis* fusion of pyran ring (C) with pyran ring (D) at C_{6a}-C_{12b}. These conclusions were further supported by the medium intensity NOE between H_{7(pro-S)}-Me_{6(pro-S)}, H_{7(pro-R)}-Me_{6(pro-S)}, and H_{12b}-Me_{6(pro-R)} and the weak NOE intensity between H₄-Me_{6(pro-R)} and H₄-Me_{6(pro-S)}, which also suggest that the six-membered pyran ring (D) adopts a twisted form. The structure was further confirmed by molecular mechanics calculations.⁷

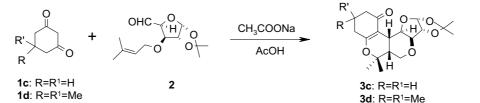
Analogous to 4-hydroxycoumarin, 4-hydroxy-1,2-dihydro-2-quinoline gave sugar fused pyrano[3,2-*c*]quinoline **3b** in 73% yield. Furthermore, cyclic 1,3-diketones such as 1,3-cyclohexadione and dimedone (Scheme 2) and 1,3dimethylbarbituric acid also afforded cycloadducts in fairly good yields (Scheme 2, Table 1, entries c, d, and e).

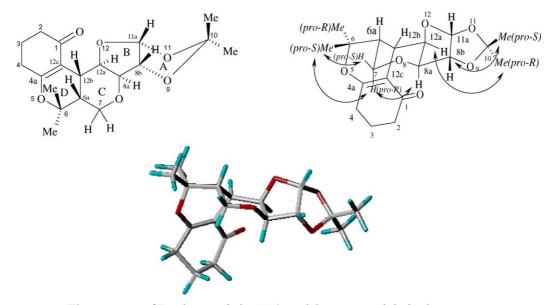
For example, in cycloadduct 3c, the pyran ring (C) exists in a chair conformation. This ${}^{6a}C_{8a}$ chair conformation is supported by NOE's between the diaxially disposed protons $H_{7(Pro-R)}$ and H_{8a} and coupling constants ${}^{3}J_{H6a-H7pro-R)} = 11.7 \text{ Hz}$, ${}^{3}J_{H6a-H7(Pro-S)} = 4.5 \text{ Hz}$ which suggest that $H_{7(Pro-R)}$ and $H_{7(Pro-S)}$ occupy axial and equatorial positions, and long range ' ω ' coupling between $H_{7(Pro-S)}-H_{12b} = 1.1 \text{ Hz}$.

Similarly, acetyl acetone and methyl acetoacetate also reacted smoothly with the sugar aldehyde to give the corresponding perhydrofuro[3,2-*b*]pyrano[4,3-*d*]pyrans (Scheme 3, Table 1, entries f and g).

We assume that the cycloaddition proceeds in a concerted manner via an *endo-E-syn* transition state. Mechanistically, a 1-oxa-1,3-butadiene may be formed from 4-hydroxycoumarin and *O*-prenylated sugar aldehyde, which can undergo an intramolecular hetero-Diels–Alder reaction leading to the *cis*-fused pyrano[3,2-*c*]coumarin derivative (Scheme 4).

In the case of unsymmetrical 1,3-dicarbonyl compounds, chemoselective synthesis of pyrano[3,2-*c*]coumarin derivatives was achieved using this procedure (Table 1, entries a, b, and g). Due to the mild basic conditions and low reactivity of the ester or amide car-



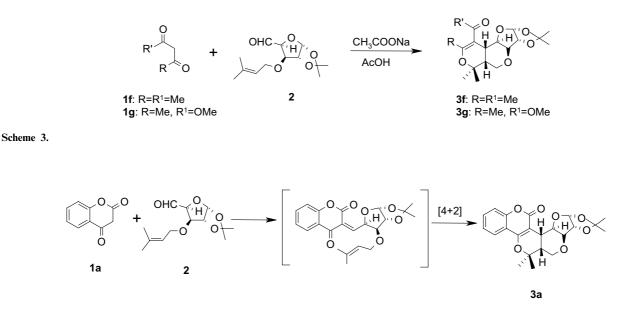


The structure of **3c**, characteristic NOE's and the energy minimised structure.

Table	1.	Domino-K	Knoevenage	hetero-D	iels–Alder	reaction

Entry	1,3-Dione	Aldehyde	Product ^a	Reaction time (h)	Yield (%) ^b
a				6.5	82
b		OHC (H) O'O'		7.5	73
с	° Co			5.0	80
d	o V O		H V H S d	6.0	82
e	Me-N O Me Ne		Me Ne ^{-N} Me ^{-N} Me ^{-N} H	8.5	72
f	Me Me		Me O O O O O O O O O O O O O O O O O O O	5.0	79
g	MeO Me		MeO Me Me V H H H H H H H H H H H H H H H H H H	6.0	70

^a All products were characterized by ¹H NMR, IR, and mass spectroscopy. ^b Isolated and unoptimized yields.



Scheme 4.

bonyl group compared to a simple carbonyl functionality, no other regioisomer was observed. Simple cyclic ketones such as cyclohexanone, cyclopentanone, and tetralone failed to give the desired products under the reaction conditions. Similarly, an *O*-allylated sugar aldehyde also did not undergo the hetero-Diels–Alder reaction with 1,3-diones. This reaction was successful only with the *O*-prenylated sugar aldehyde and 1,3dicarbonyl compounds.

In conclusion, we disclose a simple and novel protocol for the synthesis of chiral polyoxygenated heterocycles via a domino Knoevenagel-hetero-Diels–Alder reaction between an *O*-prenylated sugar aldehyde derived from D-glucose and 1,3-diones. The reaction is highly stereoselective leading to *cis*-annelated heterocycles in a singlestep operation.⁸ It is an entirely new synthetic route to construct optically active *cis*-annelated tricyclic-, tetracyclic-, and pentacyclic heterocycles.

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References and notes

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- Molecular mechanics calculations were carried out using the Sybyl 6.8 program on a Silicon Graphics O2 workstation.
- 8. General procedure: A mixture of O-prenylated sugar aldehyde 2 (1.5 mmol), 1,3-diketone 1 (1.5 mmol), and sodium acetate (4.5 mmol) in acetic acid (10 mL) was stirred at 80 °C for the appropriate time. After complete conversion, as indicated by TLC, the reaction mixture was diluted with water (10 mL) and extracted with ethyl acetate $(2 \times 15 \text{ mL})$. The combined organic layers were washed with a saturated solution of sodium bicarbonate followed by brine solution and dried over anhydrous Na₂SO₄. The resulting product was purified by column chromatography on silica gel (Merck, 100-200 mesh, ethyl acetate-hexane, 2:8) to afford the pure cis-annelated polycyclic heterocycle. Data for selected compounds, 3a: liquid, $[\alpha]_{D}^{25}$ –24.6 (c = 1.0, CHCl₃), ¹H NMR (500 MHz, CDCl₃): δ 7.72 (dd, J_{H3-H4} = 8.0, J_{H2-H4} = 1.6 Hz, 1H, H4), 7.51 (ddd, $J_{\text{H2-H3}} = 7.4$, $J_{\text{H1-H2}} = 8.3$, $J_{\text{H2-H4}} = 1.6$ Hz, 1H, H2), 7.29 (dd, $J_{\text{H1-H2}} = 8.3$, $J_{\text{H1-H3}} = 1.1$ Hz, 1H, H1), 7.23 (ddd, $J_{\text{H3}-\text{H4}} = 8.0$, $J_{\text{H2}-\text{H3}} = 7.4$, $J_{\text{H1}-\text{H3}} = 1.1$ Hz, 1H, H3), 5.84 (d, $J_{H8b-H11a} = 3.8$ Hz, 1H, H11a), 5.45 (t, $J_{\text{H8a-H12a}} = J_{\text{H12a-H12b}} = 2.3 \text{ Hz}$, 1H, H12a), 4.40 (d, $J_{\rm H8b-H11a} = 3.8 \,\rm Hz,$ 1H, H8b), 3.95 (ddd. $J_{\text{H7}(\text{pro-}S)-\text{H7}(\text{pro-}R)} = 11.7, J_{\text{H6a}-\text{H7}(\text{pro-}S)} = 4.3, J_{\text{H7}(\text{pro-}S)-\text{H12b}} =$ 1.3 Hz, 1H, $H7_{(pro-S)}$), 3.68 (d, $J_{H8a-H12a} = 2.3$ Hz, 1H, H8a), 3.45 (ddd, $J_{H6a-H12b} = 5.5$, $J_{H12a-H12b} = 2.3$, H12b), 1H, 3.17 $J_{\rm H7(pro-S)-H12b} = 1.3$ Hz, (t. $J_{\text{H7}(\text{pro-}S)-\text{H7}(\text{pro-}R)} = J_{\text{H6a}-\text{H7}\text{pro-}R)} = 11.7 \text{ Hz}, \text{ 1H, } \text{H7}_{(\text{pro-}R)},$ $J_{\text{H6a-H7(pro-}R)} = 11.7, \quad J_{\text{H6a-H7(pro-}S)} = 4.3,$ 2.33 (ddd, $J_{\text{H6a-H12b}} = 5.5 \text{ Hz}, 1\text{H}, \text{H6a}, 1.56 \text{ (s, 3H, MelO}_{(\text{pro-}R)}),$ 1.54 (s, 3H, Me6_(pro-S)), 1.37 (s, 3H, Me6_(pro-R)), 1.31 (s, 3H, Me10_(pro-S)). ¹³C NMR (75 MHz, CDCl₃): δ 161.5, 160.4, 152.7, 132.0, 123.7, 122.8, 116.4, 115.5, 112.1, 104.3, 97.6, 84.0, 79.2, 76.1, 72.5, 62.9, 35.5, 29.8, 26.8, 26.5, 25.5, 24.8.

IR (KBr): v_{max} : 2927, 1709, 1616, 1386, 1215, 1096, 1016, 763 cm⁻¹. FAB Mass: m/z: 401 (M+1), 300, 154, 137, 121, 109, 95, 83, 69, 57. **3c**: Liquid, $[\alpha]_D^{25}$ -65.5 (c = 1.0, CHCl₃), ¹H NMR (500 MHz, CDCl₃): δ 5.86 (d, $J_{H8b-H11a}$ = 3.6 Hz, 1H, H11a), 4.43 (d, $J_{H8b-H11a}$ = 3.6 Hz, 1H, H8b), 5.12 (t, $J_{H8a-H12a}$ = $J_{H12a-H12b}$ = 2.0 Hz, 1H, H12a), 3.66 (d, $J_{H8a-H12a}$ = 2.0 Hz, 1H, H8a), 3.20 (m, 1H, H12b), 3.89 (ddd, $J_{H6a-H7(Pro-R)}$ = 11.7, $J_{H6a-H7(Pro-S)}$ = $J_{H6a-H7(Pro-R)}$ = 1.7, Hz, 1H, H6a), 3.15 (t, $J_{H7(Pro-R)-H7(Pro-S)}$ = $J_{H6a-H7(Pro-R)}$ = 11.7, Hz, 1H, H7(Pro-R)), 2.15 (ddd, $J_{H7(Pro-R)-H7(Pro-S)}$ = 11.7,

 $J_{\text{H7}(\text{pro-}S)-\text{H6a}} = 4.5, J_{\text{H7}(\text{pro-}S)-\text{H12b}} = 1.1 \text{ Hz}, 1\text{H}, \text{H7}(\text{Pro-}S)), 1.55 (s, 3\text{H}, \text{Me10}(\text{pro-}R)), 1.33 (s, 3\text{H}, \text{Me6}(\text{pro-}S)), 1.31 (s, 3\text{H}, \text{Me10}(\text{pro-}S)), 1.19 (s, 3\text{H}, \text{Me6}(\text{pro-}R)), 2.46 (m, 2\text{H}, \text{H2}), 2.27 (m, 2\text{H}, \text{H4}), and 1.93 (m, 2\text{H}, \text{H3}). ^{13}\text{C} \text{NMR} (75 \text{ MHz}, \text{CDCl}_3): \delta 197.2, 171.8, 111.8, 108.3, 104.3, 83.9, 77.4, 75.9, 73.7, 62.9, 37.4, 35.5, 29.5, 28.7, 26.8, 26.5, 25.5, 24.4, 19.8. \text{IR} (\text{KBr}): v_{\text{max}}: 2932, 1735, 1647, 1601, 1380, 1295, 1218, 1145, 1094, 1015, 771 \text{ cm}^{-1}. \text{FAB Mass: } m/z: 351 (M+1), 293, 250, 233, 207, 191, 179, 165, 154, 145, 137, 123, 109, 95, 81, 69, 55.$